

Chapter 2, The Chemistry of Life (continued)**Section 2–3 Carbon Compounds (pages 44–48)**

This section explains how the element carbon is able to form millions of carbon, or organic, compounds. It also describes the four groups of organic compounds found in living things.

The Chemistry of Carbon (page 44)

1. How many valence electrons does each carbon atom have? _____

2. What gives carbon the ability to form chains that are almost unlimited in length? _____

Macromolecules (page 45)

3. Many of the molecules in living cells are so large that they are known as _____.

4. What is the process called by which macromolecules are formed? _____

5. When monomers join together, what do they form? _____

6. What are four groups of organic compounds found in living things?

a. _____

b. _____

c. _____

d. _____

Carbohydrates (pages 45–46)

7. What atoms make up carbohydrates? _____

8. Circle the letter of each sentence that is true about carbohydrates.

a. Starches and sugars are examples of carbohydrates.

b. Living things use them as their main source of energy.

c. The monomers in sugar polymers are starch molecules.

d. Plants and some animals use them for strength and rigidity.

9. Single sugar molecules are also called _____.

10. Circle the letter of each monosaccharide.

a. galactose b. glycogen c. glucose d. fructose

11. What are polysaccharides? _____

12. How do plants and animals store excess sugar? _____

Lipids (pages 46–47)

13. What kinds of atoms are lipids mostly made of? _____

14. What are three common categories of lipids?
 a. _____ b. _____ c. _____
15. Many lipids are formed when a glycerol molecule combines with compounds called _____.
16. Circle the letter of each way that fats are used in living things.
 a. As parts of biological membranes
 b. To store energy
 c. To give plants rigidity
 d. As chemical messengers
17. Complete the table about lipids.

LIPIDS

Kind of Lipid	Description
	Each carbon atom in a lipid's fatty acid chain is joined to another carbon atom by a single bond.
Unsaturated	
	A lipid's fatty acids contain more than one double bond.

Nucleic Acids (page 47)

18. Nucleic acids contain what kinds of atoms? _____

19. The monomers that make up nucleic acids are known as _____.

20. A nucleotide consists of what three parts? _____

21. What is the function of nucleic acids in living things? _____

Chapter 2, The Chemistry of Life (continued)

22. What are two kinds of nucleic acids?

a. _____

b. _____

Proteins (pages 47–48)

23. Proteins contain what kinds of atoms? _____

24. Proteins are polymers of molecules called _____.

25. What are four roles that proteins play in living things?

a. _____

b. _____

c. _____

d. _____

Reading Skill Practice

You can often increase your understanding of what you've read by making comparisons. A compare-and-contrast table helps you to do this. On a separate sheet of paper, make a table to compare the four groups of organic compounds you read about in Section 2–3. You might use the heads *Elements*, *Functions*, and *Examples* for your table. For more information about compare-and-contrast tables, see Organizing Information in Appendix A.